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THE INTRODUCTION OF A NEW SULFAMOYLATION REAGENT: N-CARBO-(TRIMETHYLSILYLOXY)SULFAMOYLCHLORIDE. VERSATILE SYNTHESES OF 3-AMINO-4-N-ALKYL AND 3-AMINO-2-N-ALKYL-5-ARYLOXY-1,2,4,6-THIATRIAZINE-1,1-DIOXIDES.

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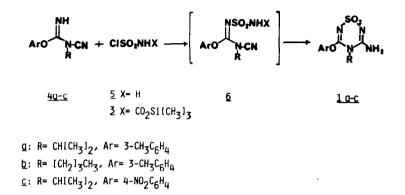
The use of a new sulfamoylation reagent, N-carbo-(trimethylsilyloxy)sulfamoyl chloride ( $\underline{3}$ ) has led to an abbreviated and high yielding preparation of 3amino-4-N-alkyl-5-aryloxy-1,2,4,6-thiatriazine-1,1-dioxides ( $\underline{1}$ ). A related approach to the 2-N-alkyl isomer ( $\underline{2}$ ) is described.

N-Alkyl-1,2,4,6-thiatriazine-1,1-dioxides are compounds of considerable interest due to their herbicidal<sup>1</sup>, fungicidal<sup>2</sup> and histamine  $H_2$ antagonist<sup>3,4</sup> activity. We wish to report the development of efficient preparations for 3-amino-4-N-alkyl-5-aryloxy-1,2,4,6-thiatriazine-1,1- dioxides (<u>1</u>, where X is <u>m</u>-cresol) and the related 2-N-alkyl isomers (<u>2</u>, where X is <u>m</u>-cresol). Furthermore, our preparation of <u>1</u> introduces the use of a readily prepared, protected derivative of sulfamoyl chloride: N-carbo-(trimethylsilyloxy)sulfamoyl chloride(<u>3</u>). This reagent should be of general utility in the preparation of a wide variety of S and N containing heterocycles.

Our methodology has proved to be advantageous for the synthesis of both 4-N-alkyl and 2-N-alkyl isomers substituted with sterically demanding

CISO2NHCO2SiMe3 3 2

alkyl groups. In work reported previously<sup>5</sup>, the 4-N-isopropyl derivative of 1 (where X is methylthic) could not be prepared<sup>5a</sup> and the 2-N-isopropyl derivative of 2 (where X is methylthic) was available in low yield  $(6\%)^{5b}$ . We have obtained yields of 81% and 54%, respectively for the corresponding isopropyl derivatives of 1 and 2 (where X is m-cresol).



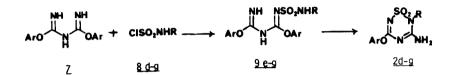
The scheme for the synthesis of <u>1</u> envisioned the cyclization of readily available N-cyano-N-alkyl pseudoureas  $\frac{4a-c}{6}^{6}$  with sulfamoyl chloride (<u>5</u>). However, we found the preparation of <u>5</u> (by the addition of one equivalent of water to chlorosulfonyl isocyanate<sup>7</sup>) to be inconvenient on a small scale and the purification of bifunctional <u>5</u> by distillation resulted in a low yield. Our solution to this problem was the development of a convenient substitute for <u>5</u>, N-carbo(trimethylsilyloxy)sulfamoyl chloride (<u>3</u>).

Reagent 3 was available from the addition of one equivalent of 2-(trimethylsilyl)ethanol to chlorosulfonyl isocyanate in a variety of solvent mixtures (carbon tetrachloride, 4:1 carbon tetrachloride:dichloromethane and 4:1 carbon tetrachloride:cyclohexane). The initial adduct was detected by <sup>1</sup>H NMR<sup>8</sup>, but during the removal of solvent under the usual conditions (~20 mm Hg, 40<sup>°</sup>) the elimination of ethylene occurred to give  $3^9$ .

$$Clso_NCO + HO(CH_2)_SIMe_3 \longrightarrow Clso_NHCO_2(CH_2)_SIMe_3 \xrightarrow{-CH_2CH_2} Clso_NHCO_2SIMe_3$$

<u>3</u>

Sulfamoylation of  $\underline{4a}$  and  $\underline{4b}$  with a mixture of one equivalent each of freshly prepared 3 and triethylamine in dichloromethane gave 6. Compound 6 was deprotected and cyclized in <u>situ</u> by the addition of an excess of water with vigorous stirring. Heterocycles <u>1a</u> and <u>1b</u><sup>10</sup> precipitated from the reaction mixture in excellent yield (see Table). Due to the decomposition of <u>4c</u> by triethylamine, <u>1c</u> was prepared by the addition of 0.5 equivalent of 3 to <u>4c</u> and the product isolated by column chromatography after hydrolysis.



Ar= 3-CH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>, R=  $\underline{d}$ : CH<sub>3</sub>,  $\underline{e}$ : CH[CH<sub>3</sub>]<sub>2</sub>,  $\underline{f}$ : [CH<sub>2</sub>]<sub>3</sub>CH<sub>3</sub>,  $\underline{g}$ : C<sub>6</sub>H<sub>11</sub>

The preparation of 2 followed in analogous fashion from imino bis[(3methylphenyl)carbimidic acid] (7)<sup>11</sup>. Sulfamoylation of 7 with one equivalent of N-methylsulfamoyl chloride (8d)<sup>12</sup> and triethylamine in tetrahydrofuran gave 2d directly in quantitative yield. Similar treatment of 7 with the higher alkyl derivatives  $\underline{8e-g}^{12}$  gave only the adducts  $\underline{9e-g}$ . These derivatives of 9 were cyclized with two equivalents of sodium hydride in tetrahydrofuran at room temperature to heterocycles  $\underline{2e-g}$  in good yield (see Table). The reaction time for the conversion of 9 to 2 varied directly with the increasing steric bulk of the N-alkyl group<sup>13</sup>. A similar steric inhibition was observed in cyclizations with N-alkylsulfamoyl derivatives of isothiobiuret similar to 9 (where ArO was the alkylthio group)<sup>5b</sup>.

т	а	b	l	е
-	_	_	-	-

		R	Ar	mp	<u>Yield (%)</u>
1	а	<sup>CH(CH</sup> 3)2	<sup>3-CH</sup> 3 <sup>C</sup> 6 <sup>H</sup> 4	>250°	86
	b	(CH <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>	3-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	208–213 <sup>0</sup>	81
	с	CH(CH <sub>3</sub> )2	4-N02 <sup>C6H4</sup>	>250 <sup>0</sup>	42
2	d	СНЗ	3-CH3C6H4	>250 <sup>0</sup>	100
	е	CH(CH <sub>3</sub> ) <sub>2</sub>	3-CH3C6H4	190-191 <sup>0</sup>	54
	f	(CH <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>	3-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	171-173 <sup>0</sup>	80
	g	C <sub>6</sub> H <sub>11</sub>	3-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	216-217 <sup>0</sup>	35
2	е	CH(CH <sub>3</sub> ) <sub>2</sub>	3-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	107-110 <sup>0</sup>	80
	f	(CH <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>	3-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	100 <b>-</b> 102 <sup>0</sup>	98
	g	<sup>C</sup> 6 <sup>H</sup> 11	3-CH3C6H4	141-143 <sup>0</sup>	73

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- 8. 90 MHz <sup>1</sup>H NMR (CCl<sub>4</sub>, CH<sub>2</sub>Cl<sub>2</sub> standard)  $\delta$ : 8.8 (s, 1H), 4.3 (t, J=8Hz, 2H), 1.15 (t, J=8Hz, 2H) and 0.1 (s, 9H).
- 9. Reagent 3: One equivalent of 2-(trimethylsilyl)ethanol was added dropwise to a 2M solution of chlorosulfonyl isocyanate in 4:1 carbontetrachloride:hexane at ambient temperature. During the addition the reaction temperature rose from 25 to 40°C and the mixture was allowed to stir for 1h. After removal of the solvent, 3 was isolated in 100% yield as a low melting solid (m.p. <40°) and was used directly in sulfamoylations. 90MHz <sup>1</sup>H NMR (CCl4, CH<sub>2</sub>Cl<sub>2</sub> standard) &: 8.6 (s, 1H) and 0.1 (s, 9H);IR 1740 (s), 1720 (s), 1410 (s), 1350 (s), 1160 (s) and 1050 (s).
- Compounds <u>1a-c</u>, <u>2d-g</u> and <u>9e-g</u> gave satisfactory combustion analyses. The <sup>1</sup>H NMR, infrared and mass spectra were in agreement with their proposed structures.
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- The approximate times for the complete conversion of <u>9</u> to <u>2</u> were for <u>9e</u>: 1h, <u>9f</u>: 0.25h, <u>9h</u>: 3h. Reactions were monitored by TLC.

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